

Meteorology

- Simple meteorological models
 - Obtain concentration of species
 - Mass balance equation
 - Box model
 - Puff models

- Winds
 - Transport in the atmosphere
 - Forces affecting the wind
 - Flow patterns around high and low pressures
 - Atmospheric stability
 - General circulation → climate zones

Literature connected with today's lecture:

Jacob, chapter 3 - 4

Exercises:

3:1 – 3:5; 4:1 – 4:5

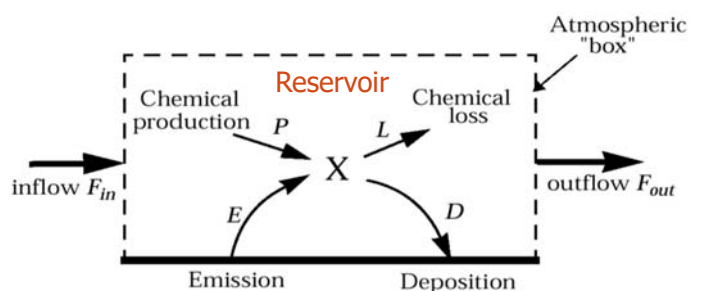
Simple Meteorological Models

Atmospheric concentration is controlled by:

- **Emission (E):** natural and anthropogenic sources
- **Deposition (D):** dry and wet deposition
- **Transformation (P,L):** chemical reactions and phase transitions
- **Transport (F):** winds

Box model (Eulerian)

- Imaginary box in the atmosphere
- Homogeneous concentration of X in the box is assumed in the model
 - Amount of species X
 - Sources: E, P, F_{in}
 - Sinks: D, L, F_{out}
 - Dimension: mass/time, with X amount as mass



Box Model – The Mass Balance Equation

$$dm/dt = \Sigma_{\text{sources}} - \Sigma_{\text{sinks}} = F_{\text{in}} + E + P - F_{\text{out}} - D - L$$

- Residence time in a box
 - Average time a species spends in the box
 - $\tau = m/(F_{\text{out}} + L + D)$ ([mass in the box]/[outflow])
- First order sinks
 - F_{out} , L and D often proportional to the amount in the box
 - $F_{\text{out}} + L + D = (k_{\text{out}} + k_l + k_d)m = km$
 - Loss rate constant: $k = (F_{\text{out}} + L + D)/m = 1/\tau$

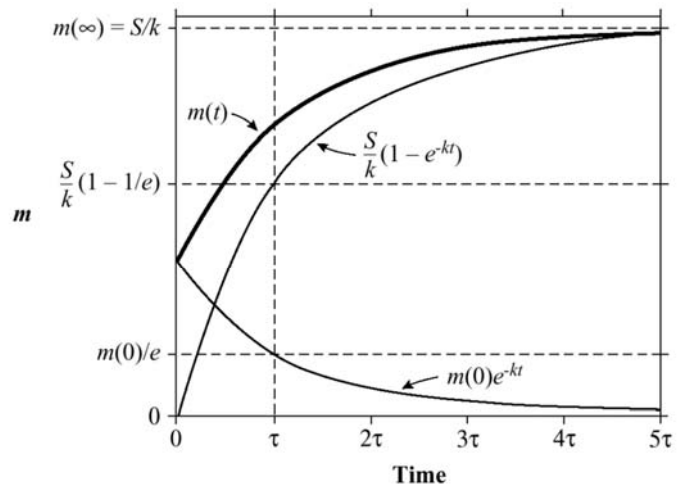
Box Model – Common Special Case

- Sources independent of m
- Sinks proportional to m

Mass balance equation:

$$\begin{aligned} \frac{dm}{dt} &= \sum \text{sources} - \sum \text{sinks} = \\ &= F_{in} + E + P - F_{out} - D - L \end{aligned}$$

$$\begin{aligned} F_{in} + E + P &= S \quad (S \text{ indep. of } m) \\ F_{out} + D + L &= km \quad (\text{sinks prop. } m) \\ \frac{dm}{dt} &= S - km \\ \text{Compute } m(t)! \end{aligned}$$



$$\begin{aligned} \int \frac{dm}{S - km} &= \int dt \Rightarrow \\ m(t) &= m(0)e^{-kt} + \frac{S}{k}(1 - e^{-kt}) \\ t \rightarrow \infty &\Rightarrow m \rightarrow S/k \end{aligned}$$

Exercise 3-3 in Jacob: The sink of CFC-12 (CF_2Cl_2) is exclusively photolysis (residence time 100 years). Year 1980 the concentration was 400 pptv and the rate of increase 4% per year. Calculate the 1980 CFC-12 emission!

Use the mass balance equation:

$$\frac{dm}{dt} = F_{in} + E + P - F_{out} - L - D$$

Box: Entire atmosphere $\Rightarrow F_{in} = F_{out} = 0$
 $L + D = L = km$ (sink: photolysis only)
 $E + P = E$ (no chemical production)

$$\left. \begin{array}{l} \\ \\ \end{array} \right\} \frac{dm}{dt} = E - km$$

dm/dt is given by a relative measure: $k_s = 4\%$ per year): $\frac{dm}{dt} = k_s \times m$

Sink (photolysis): $k = 1/\tau = 0.01 \text{ year}^{-1}$

Emissions (from mass balance equation): $E = \frac{dm}{dt} + km = (k_s + k)m$

Enter numbers: $E = (0.04 \text{ year}^{-1} + 0.01 \text{ year}^{-1}) m$

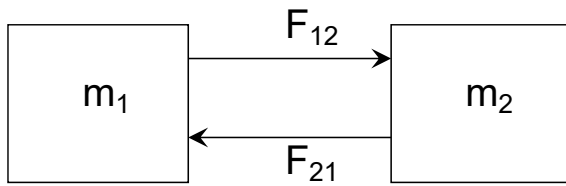
$$m = M_{CFC} n_{CFC} = \left[C_{CFC} = \frac{n_{CFC}}{n_a} \right] = M_{CFC} C_{CFC} n_a = M_{CFC} C_{CFC} \frac{m_a}{M_a} = M_{CFC} C_{CFC} \frac{4\pi R^2 P}{M_a g}$$

Result: $E = 4.4 \times 10^8 \text{ kg/year}$

Multiple Box Models

- A single box model sometimes oversimplifies a problem
- Multi-box models allow concentration to vary between boxes

Two-box Model



- Mass balance equation for box 1:

$$\frac{dm_1}{dt} = E_1 + P_1 - L_1 - D_1 - F_{12} + F_{21}$$

- First order process:

- $F_{12} = k_{12}m_1$

- $F_{21} = k_{21}m_2$

- Coupled differential equations:

$$\frac{dm_1}{dt} = E_1 + P_1 - L_1 - D_1 - k_{12}m_1 + k_{21}m_2$$

$$\frac{dm_2}{dt} = E_2 + P_2 - L_2 - D_2 + k_{12}m_1 - k_{21}m_2$$

Lagrangean Model

- **Box model:** Air flows in and out of the box (Eulerian model)
- **Puff models:** Follows an air parcel in the atmosphere (Lagrangean model)

- Mass bal. eqn. **Lagrangean model:**

- $d[X]/dt = E + P - D - L$

- Advantage: $F_{in} = F_{out} = 0$

- Disadvantage: Limited range due to turbulence that diffuses the air parcel

- Common applications:

- Smoke plumes

- $d[X]/dt = E + P - D - L - k_{dil}([X] - [X]_b)$

- k_{dil} = dilution constant

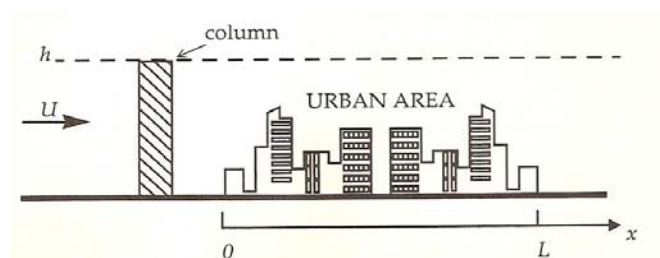
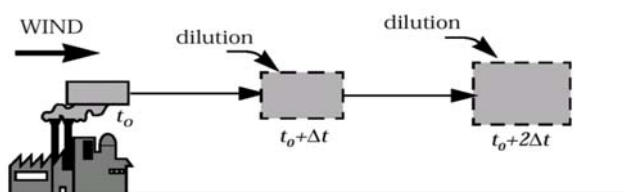
- Column model

- $d[X]/dt = E/h + P - D - L$

- E = emission per area and time unit

- h = height of the column (mixing height)

Smoke plume



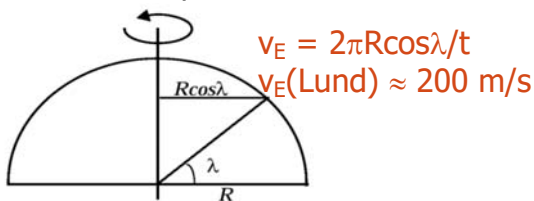
Transport in the Atmosphere

Forces that affect winds:

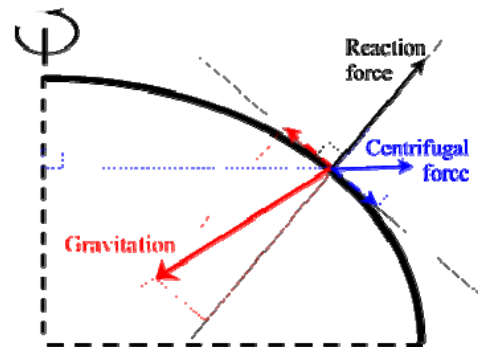
- **Gradient force** - Horizontal pressure gradient, strong winds (high-, low pressure)
- **Gravitational force** - induces vertical air motions related to density
- **Coriolis force** – Caused by the rotation of the earth
- **Friction force** – Acts on winds in contact with the ground

Coriolis Force

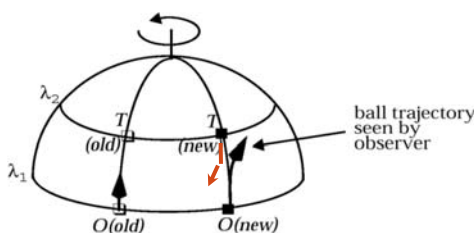
- Rotation of the earth
 - An object at the surface has velocity v_E
 - v_E depends on latitude



- Coriolis force along latitude



- Coriolis force along longitude



Rest: Horizontal comp. cancel out
 \Rightarrow The earth flattened at the poles

Motion west - east at northern hemisphere:
 \Rightarrow increased velocity \Rightarrow increased centrifugal force \Rightarrow
 Bends towards the equator (right)
 East - west \Rightarrow bends towards the north pole (right)
 Southern hemisphere: bends to the left

Winds both along longitude and latitude bends
 Northern hemisphere: Bends to the right Southern hemisphere: To the left

Coriolis Force

- Coriolis acceleration

$$\gamma_c = 2\omega v \sin\lambda$$

$\omega = 2\pi/T$ (angular velocity)

$v =$ velocity relative the earth

$\Delta X =$ distance

$\lambda =$ latitude

- Resulting displacement

$$\Delta Y = \omega(\Delta X)^2 \sin\lambda / v$$

Example ($\lambda = 42^\circ$):

1. Snow ball $v = 20$ km/h

$\Delta X = 10$ m:

$\Delta Y = 1$ mm

2. Missile $v = 2000$ km/h

$\Delta X = 1000$ km:

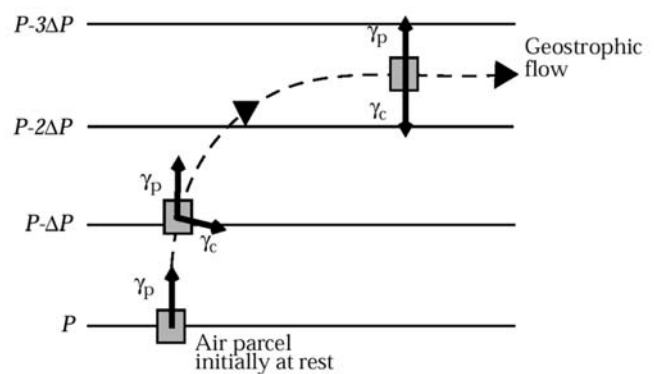
$\Delta Y = 100$ km

Geostrophic Wind

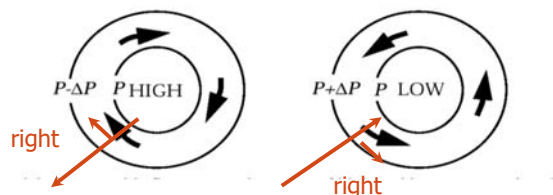
- Pressure gradient induces a wind

Motion \Rightarrow Coriolis force

- Balance between gradient and Coriolis forces \Rightarrow
- **Geostrophic wind** – Wind along isobar \Rightarrow
- No air transport to/from low/high pressure!

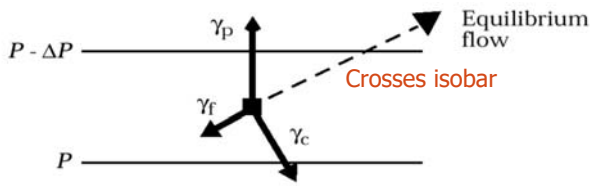


- Geostrophic wind around high and low pressures (northern hemisphere)



Friction Force

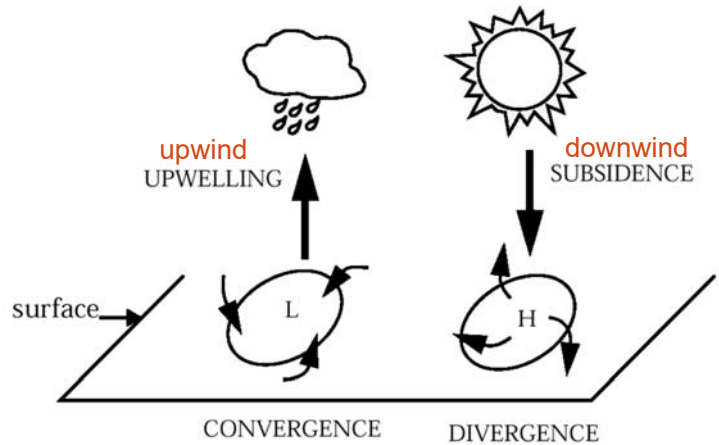
- Effect of friction against the ground



- Friction acts against the motion and reduces the speed
- \Rightarrow reduced Coriolis force
- \Rightarrow wind component crosses isobars close to the ground

- High and low pressures

- Wind to low pressures close to the ground
 - Convergence - Upwind in low pressures \Rightarrow
 - Air expands and cool \Rightarrow Clouds
- Wind from high pressure close to the ground
 - Divergence - Subsidence in high pressures \Rightarrow
 - Air compressed - warmed \Rightarrow Clear weather



Vertical Transport

- **A fluid at equilibrium:** The force from a pressure gradient acting on a volume element is balanced by the gravitational force
 - Gradient force: $F_g = \rho' Vg$
- Convection (buoyancy)
 - Caused by difference in density :

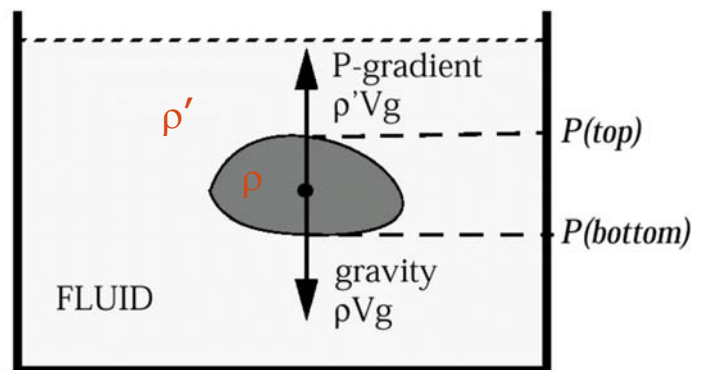
- Gravitational force: $mg = -\rho Vg$

- Resulting force:

$$F = (\rho' - \rho)Vg$$

Buoyant acceleration:

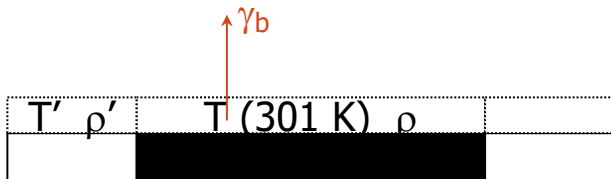
$$\gamma_b = F/m = (\rho' - \rho)g/\rho$$



Exercise 4-1 (Jacob)

Assume that the air surface temperature over a black parking lot is 1 K higher than in the surrounding air (300 K).

Calculate the buoyant acceleration!



$$\text{Buoyant acceleration: } \gamma_b = (\rho' - \rho)g / \rho$$

$$\text{From the ideal gas law: } \rho = PM_a / RT$$

$$\gamma_b = (T - T')g / T'$$

$$\gamma_b = 0.033 \text{ m/s}^2$$

Calculate the upward wind speed after 1 s?

$$w = t \gamma_b = 3.3 \text{ cm/s}$$

Comparison:

Global circulation: $w \approx 0.1 \text{ cm/s}$

Cumulus clouds: w several m/s

Does the velocity continue to increase? →

Atmospheric stability

Adiabatic Lapse Rate

- Small heat exchange in vertical transport ($dQ \approx 0$) ⇒ Adiabatic process
 - Upwards ⇒ expansion ⇒ cooling
 - Downwards ⇒ compression ⇒ warming

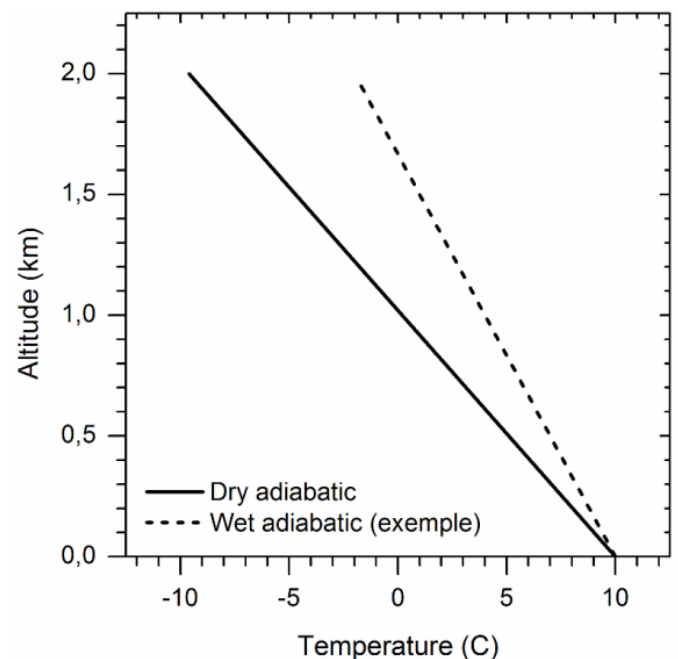
- **Dry adiabatic lapse rate**

Can be shown (see the text book):

$$\Gamma = -dT/dz = g/C_p = 9.8 \text{ K/km}$$

- **Wet adiabatic lapse rate**

Cloud formation ⇒ heat of condensation released ⇒ weaker gradient; Γ_w 2-7 K/km

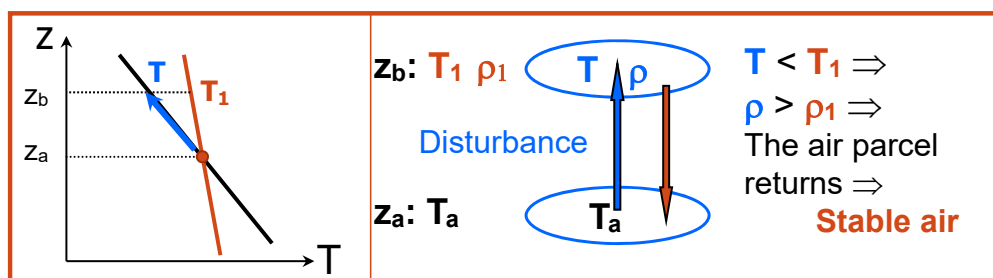


Air masses in vertical motion follow the adiabatic lapse rate (dry or wet)

Atmospheric Stability

Sometimes strong vertical motions and turbulence – sometimes not – WHY?

Disturbance: An mass air at height z_a is lifted to $z_b \Rightarrow$ Expands and cools adiabatically to T



- Atm. Lapse rate 1
- Dry adiabatic lapse rate
- Atm. Lapse rate 2

At z_a (initially): T_a

At z_b :

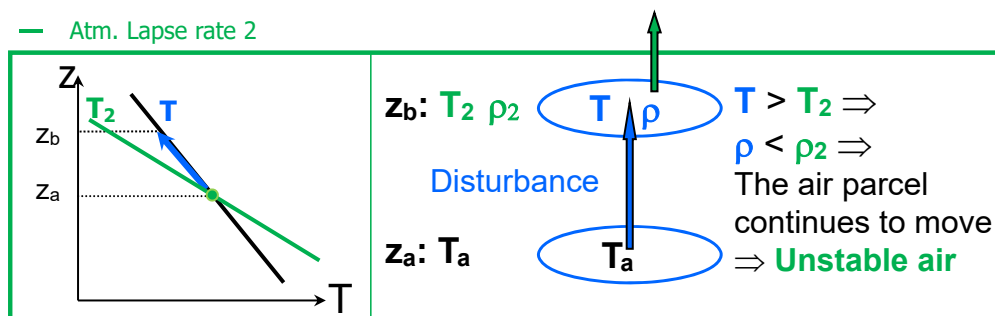
Disturbed air mass:

$$T = T_a - \Gamma(z_b - z_a)$$

Surrounding atmosphere:

Case 1: T_1

Case 2: T_2



Atmospheric Stability (No cloud)

- Example 1:

- The atmospheric temperature decreases by 15 °C over 1 km in altitude. Stable or unstable?

- $dT_{\text{atm}}/dz = -15/1 = -15 \text{ K/km} < -\Gamma$

- Decreases faster than the dry adiabatic lapse rate ($\Gamma = 9.8 \text{ K/km}$) => **unstable** => **turbulence, vertical winds**

- Example 2:

- The atmospheric temperature increases by 15 °C over 1 km in altitude. Stable or unstable?

- $dT_{\text{atm}}/dz = +15/1 = 15 \text{ K/km} > -\Gamma$

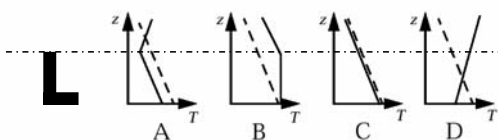
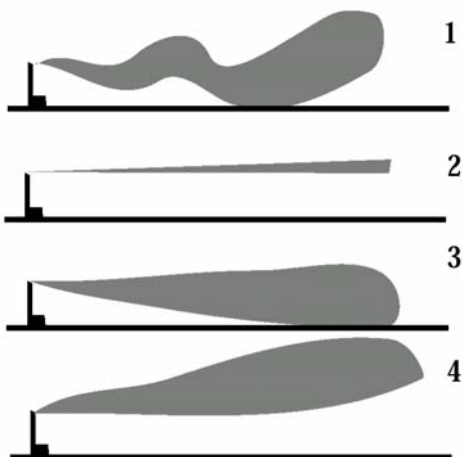
- **Stable** => **no vertical winds**

- Increased temperature with altitude: **Inversion**, extremely stable

Stable air: The atmospheric lapse rate **smaller** than the adiabatic lapse rate
Unstable air: The atmospheric lapse rate **larger** than the adiabatic lapse rate
Neutral air: $dT_{\text{atm}}/dz = -\Gamma$ is **unstable**

Exercise 4:1

- Match each picture to the right diagram!

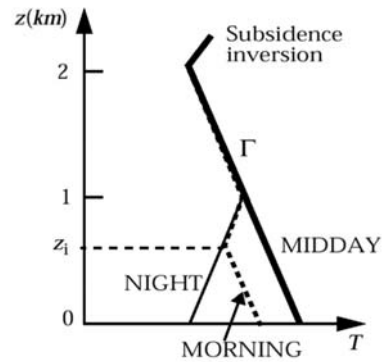


— Atmosphere
 - - Adiabatic lapse rate

- Neutral: $-dT_{\text{atm}}/dz = \Gamma$; => vertical motions (like unstable air)
- **A:** Neutral at low levels, inversion at high levels. Air can move downwards – **3**
- **B:** Stable at low levels, neutral at high levels – **4**
- **C:** Neutral up- and downwards – **1**
- **D:** Inversion. Air cannot move vertically - **2**

Tropospheric Lapse Rate

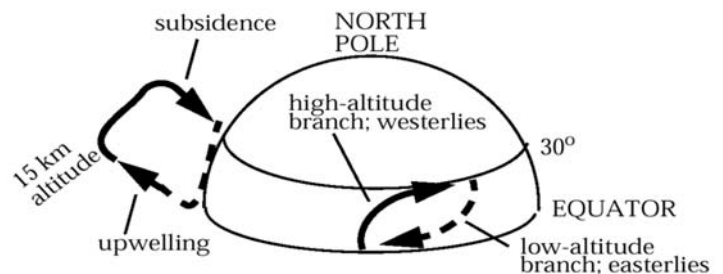
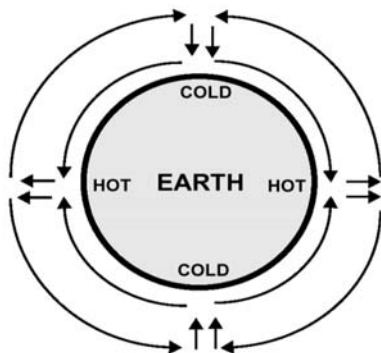
- The troposphere on average stable
 - Lapse rate 6.5 K/km
 - Not the adiabatic lapse rate because
 - heat of condensation from cloud formation
 - radiation
- Heating/cooling from the ground affect atmospheric stability
 - Night-time heat radiation cools the ground and the lowest atmosphere \Rightarrow stability
 - Often night-time inversion
 - The sun heats the ground and the lowest atmosphere \Rightarrow unstable
 - Vertical motion drives air parcels to the level of adiabatic equilibrium (**neutral**)



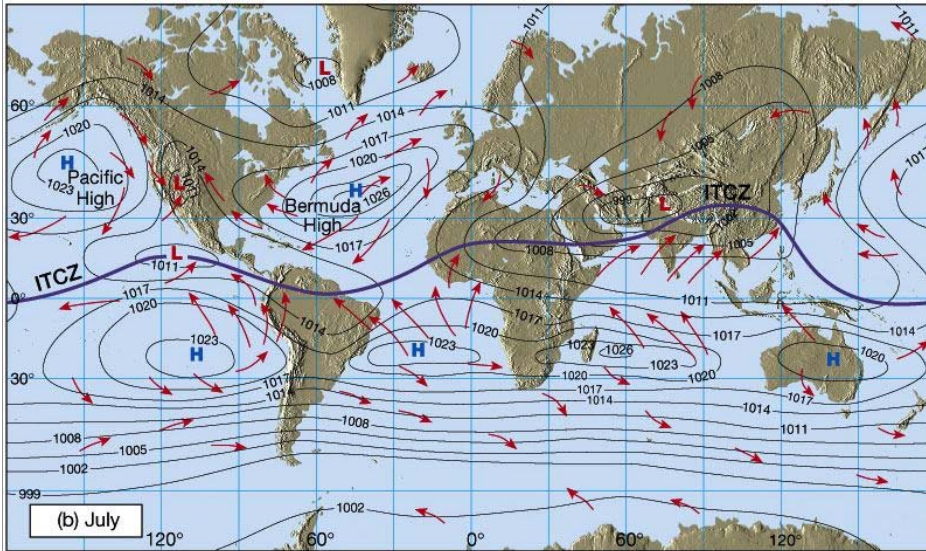
- Mixing layer
 - Continual vertical motions caused by instability
 - The height of the mixing layer varies over the day
 - The strong turbulence causes efficient transport within the mixing layer

Global Circulation

- Old model - global sea-breeze
- Coriolis force prevents equator-to-pole circulation
- The Hadley cell



Global Circulation



The Hadley cell:

- Inter-tropical convergence zone (ITCZ)
 - Strong upwind
 - Clouds and precipitation
- Subtropical high pressures (30° N,S)
 - Downwind => Dry

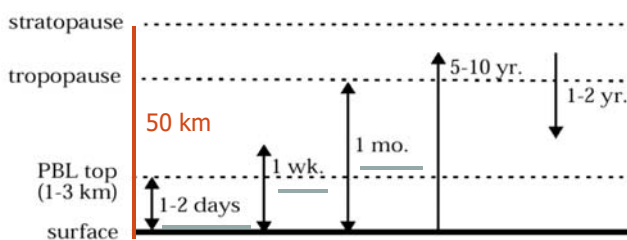
Surface winds:

- Easterly both sides of the ITCZ
- Mainly westerly at higher latitudes
 - More pronounced at SH, due to less land => less friction force => less disturbance of the geostrophic wind

Time Scales of Atmospheric Transport

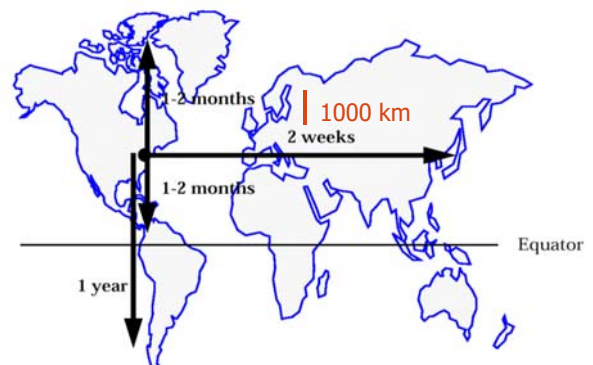
Vertically:

- Turbulent transport in the troposphere
- $\Delta t = (\Delta z)^2 / (2K_z)$
 - K_z turbulent "diffusion coeff." (empirical)
 - $\langle K_z \rangle = 2 \cdot 10^5 \text{ cm}^2/\text{s}$ in troposphere
- Transport from the stratosphere faster (residence time shorter) due to $m_s \ll m_T$



Horizontally:

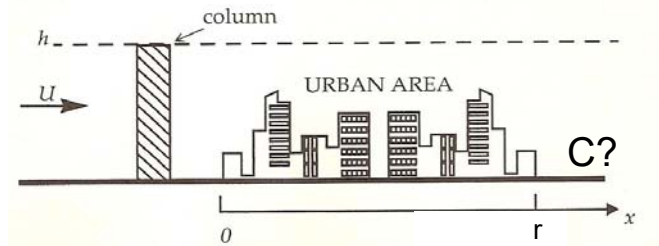
- North – south within hemisphere: 1- 2 months
- Between north – south hemispheres: 1 year
- East – west much faster than north – south
 - Caused by geostrophic flow induced by latitude-dependent temperature



Example Lagrangean Model: A village emits $3 \mu\text{g SO}_2$ per m^2 and s. A 1.5 m/s wind blows over the village that extends 3 km in the wind direction. Air is mixed up to 1 km altitude. Calculate the SO_2 concentration added to the air by the village. (Assume P, L, D are 0)

$$E = 3 \mu\text{g SO}_2 \text{ per m}^2 \text{ and s}$$

$$u = 1.5 \text{ m/s}; r = 3 \text{ km}; h = 1 \text{ km}$$



$$\text{Mass balance eqn: } dC/dt = E/h + P - L - D = E/h \quad (1)$$

$$dC/dt = dC/dx \cdot dx/dt = u \cdot dC/dx \quad (2)$$

$$\text{Combine (1) and (2): } dC/dx = E/uh$$

$$\text{Integrate } dC/dx: C(r) = [r - 0]E/uh = Er/uh = 3 \cdot 3000 / (1.5 \cdot 1000) = 6 \mu\text{g/m}^3$$

The village increased the SO_2 concentration by $6 \mu\text{g/m}^3$